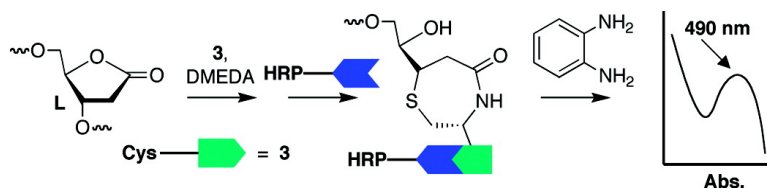


## Selective Detection of 2-Deoxyribonolactone in DNA

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*J. Am. Chem. Soc.*, **2005**, 127 (9), 2806-2807 • DOI: 10.1021/ja0426185 • Publication Date (Web): 10 February 2005

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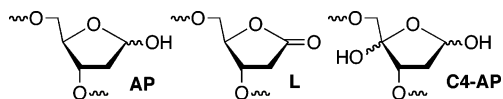
## Selective Detection of 2-Deoxyribonolactone in DNA

Kousuke Sato and Marc M. Greenberg\*

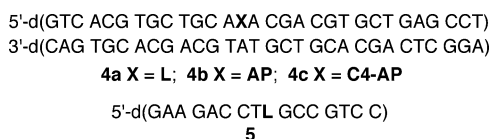
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DNA lesions give rise to mutations and can be a primary chemical transformation en route to cancer. Elegant methods exist for detecting specific sequences in DNA, which are useful for sensing mutations and single nucleotide polymorphisms.<sup>1–4</sup> However, sensitive detection of DNA lesions is mostly limited to mass spectrometric analysis of individual lesions (or their nucleobases) following complete chemical or enzyme degradation of the nucleic acids.<sup>5,6</sup> Appropriately tagged alkoxy amines (e.g., aldehyde reactive probe, **ARP**), which selectively react with abasic sites (**AP**), are one of the few reagents available for detecting specific lesions in DNA.<sup>7,8</sup> We now report a sensitive method for selectively detecting the oxidized abasic lesion 2-deoxyribonolactone (**L**) in DNA.

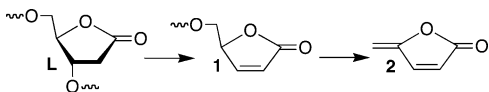


2-Deoxyribonolactone is produced in DNA by the neocarzinostatin chromophore, and other damaging agents, including  $\gamma$ -radiolysis.<sup>9,10</sup> Recent experiments using independently generated reactive intermediates suggest that **L** is produced in greater quantities than predicted by the accessibility of the C1'-hydrogen atom to a diffusible species.<sup>11–13</sup> The formation and detection of **L** is important because the lesion possesses interesting properties. For example, 2-deoxyribonolactone is one of only two lesions known to form DNA-protein cross-links.<sup>14–16</sup> Furthermore, **L** exhibits a mutation spectrum in *Escherichia coli* that is clearly distinct from that of an **AP** site.<sup>17</sup>

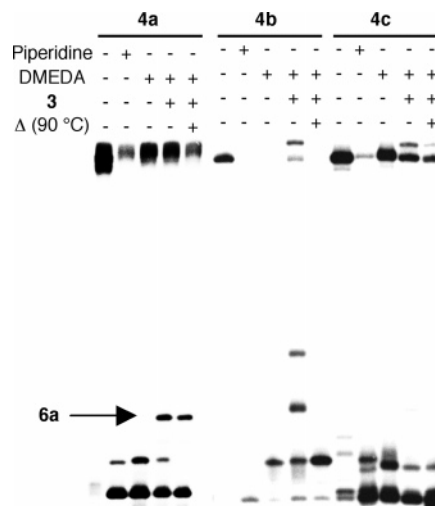


2-Deoxyribonolactone has been detected using GC/MS and/or HPLC following alkaline digestion of oxidatively damaged DNA as its persilylated ether and by direct analysis of its elimination product (**2**).<sup>18,19</sup> These methods underestimate the amount of **L** present in DNA because the conditions used to release the lactone prior to silylation may destroy some of the lesion, and **2** is unstable (Scheme 1). We designed a strategy that takes advantage of the

### Scheme 1

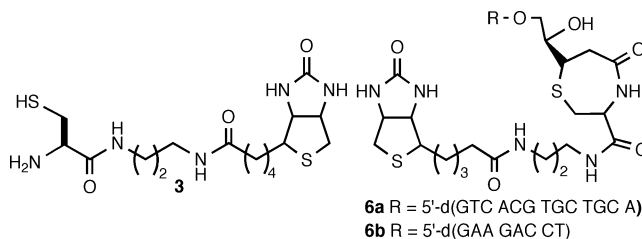


unique reactivity of **L** and incorporates an amplification mechanism to increase sensitivity. The butenolide (**1**) has previously been generated using *N,N'*-dimethylethylenediamine (DMEDA) and trapped in situ by nucleophiles.<sup>20</sup> The reaction products have been used as a fingerprint to detect **L** in oligonucleotides.<sup>12,21</sup> These



**Figure 1.** Autoradiogram demonstrating selective detection of **L** by **3** (50 mM) in the presence of DMEDA (50 mM).

studies suggested that an appropriately substituted cysteine could serve as a sensor for 2-deoxyribonolactone by trapping **1**. The utility of biotinylated cysteine (**3**) for detecting **L** was examined using a duplex (**4a**) containing the lesion at a defined site.<sup>22</sup> A cleavage product (**6a**) with considerably retarded migratory ability relative to those consisting of 3'-phosphate or amine adducts was observed when **4a** was treated with **3** and DMEDA (Figure 1). This product was stable to heating at 90 °C, which is consistent with the expectations for cyclic amide **6a**. Furthermore, MALDI-TOF MS analysis of reaction between **5** and **3** under comparable conditions detected a product that corresponds to the expected molecular weight of **6b**.<sup>23</sup>

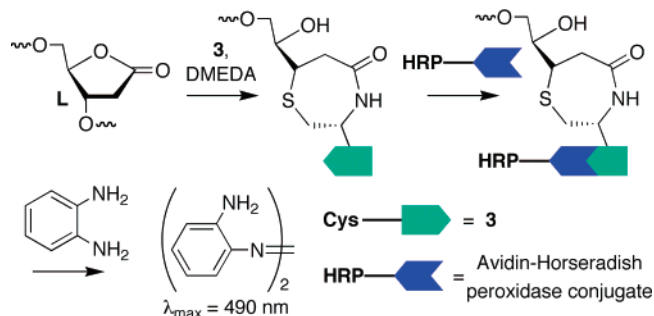


Optimized reaction conditions that produce **6a** in 20% yield were used to determine the selectivity of **1** for the butenolide derived from **L** over **AP** and **C4-AP**, which were independently generated in otherwise identical duplexes (**4b**, **4c**).<sup>24,25</sup> The latter is formed in significant amounts by agents such as bleomycin and the enediyne antitumor antibiotics, whereas endogenous conditions produce approximately 10,000 **AP** sites per cell per day.<sup>9,26,27</sup> Cysteine derivative **3** did not trap the **C4-AP** lesion in **4c** (Figure 1). In contrast, two adducts were detected when the **AP** site (**4b**) was subjected to the conditions used for trapping **L**, but these were

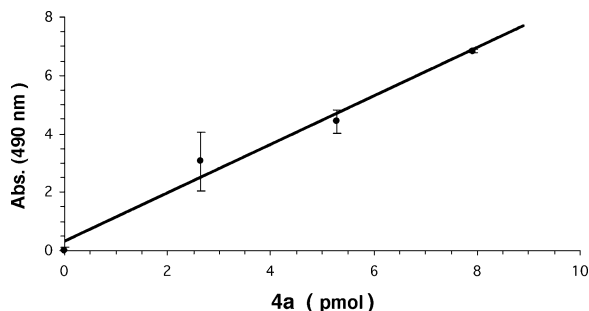
unstable at 90 °C. These data show that **3** selectively detects **L** over **AP** or **C4-AP**. The biotinylated cysteine also does not react with thymidine glycol, a prototypical dihydropyrimidine lesion.<sup>23</sup> Finally, it is worth noting that **L** (**4a**) does not react with commercially available **ARP** (data not shown), suggesting that simultaneous detection of **AP** and 2-deoxyribonolactone lesions will be possible using these two reagents.

The sensitivity for 2-deoxyribonolactone (**L**) detection was enhanced by taking advantage of biotin–avidin binding in conjunction with horseradish peroxidase (HRP) catalysis (Scheme 2). After

Scheme 2



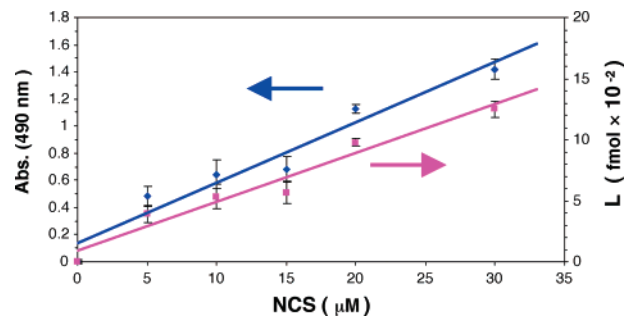
treating **4a** with DMEDA and **3** (followed by heating at 90 °C) and removing excess reagents, the DNA was adsorbed on the surface of a well in a microtiter plate. The plate was washed with avidin–HRP conjugate, followed by addition of 1,2-diaminobenzene. Oxidation of the diamine was monitored at 490 nm after 15 min (Figure 2). A linear correlation with respect to the quantity of **L** was observed, and absolute amounts of 2-deoxyribonolactone present were determined via phosphorimage analysis of radiolabeled **4a**.



**Figure 2.** Detection of 2-deoxyribonolactone (**L**) in **4a** by **3** and avidin–HRP conjugate.

Finally, the horseradish peroxidase assay was used to measure **L** in DNA treated with the neocarzinostatin chromophore (NCS). A PCR fragment (287 nt) produced from amplification of pM13 was reacted with varying concentrations of NCS. The oxidatively damaged DNA was then subjected to reaction with **3**, followed by amplification of the signal (Figure 3). The absorbance at 490 nm was corrected for the yield of biotinylated adduct (20%) and calibrated using an identical duplex in which one of the primers contained a 5′-biotin. The lowest concentration of NCS (5 μM) employed produced 390 fmol of **L**. No signal above background was observed when the PCR fragment was treated with acid or with bleomycin, conditions that result in **AP** and **C4-AP** lesions, respectively.<sup>23</sup> However, abasic sites were detected following treatment with **ARP**.<sup>23</sup>

In conclusion, we have developed the first method to selectively detect the oxidized abasic lesion, 2-deoxyribonolactone (**L**), by



**Figure 3.** Detection of 2-deoxyribonolactone (**L**) by **3** and avidin–HRP conjugate in duplex DNA (287 nt) treated with varying amounts of NCS.

taking advantage of its distinctive chemical reactivity. The cysteine biotin probe (**3**) will be useful for measuring the amounts of 2-deoxyribonolactone formed in DNA exposed to various forms of oxidative stress.

**Acknowledgment.** We are grateful for generous support from the National Institute of General Medical Sciences (GM-054996, GM-063028) and Professor Irving Goldberg (Harvard University) for providing NCS. Dedicated to Professor Peter Dervan on the occasion of his 60th birthday.

**Supporting Information Available:** Experimental procedures for carrying out lactone detection using **3** and control experiments; MALDI-TOF analysis of reaction of **5**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

## References

- Tyagi, S.; Marras, S. A. E.; Kramer, F. R. *Nat. Biotechnol.* **2000**, *18*, 1191–1196.
- Okamoto, A.; Tanaka, K.; Fukuta, T.; Saito, I. *J. Am. Chem. Soc.* **2003**, *125*, 9296–9297.
- Sando, S.; Abe, H.; Kool, E. T. *J. Am. Chem. Soc.* **2004**, *126*, 1081–1087.
- Park, S.-J.; Taton, T. A.; Mirkin, C. A. *Science* **2002**, *295*, 1503–1506.
- Dizdaroglu, M.; Jaruga, P.; Birincioglu, M.; Rodriguez, H. *Free Rad. Biol. Med.* **2002**, *32*, 1102–1115.
- Cadet, J.; Douki, T.; Gasparutto, D.; Ravanat, J. L. *Mutat. Res.* **2003**, *531*, 5–23.
- Ide, H.; Akamatsu, K.; Kimura, Y.; Michiue, K.; Makino, K.; Asaeda, A.; Takamori, Y.; Kubo, K. *Biochemistry* **1993**, *32*, 8276–8283.
- Lhomme, J.; Constant, J. F.; Demeunynck, M. *Biopolymers* **1999**, *52*, 65–83.
- Xi, Z.; Goldberg, I. H. In *Comprehensive Natural Products Chemistry*; Kool, E. T., Ed.; Elsevier: Amsterdam, 1999; Vol. 7, pp 553–592.
- von Sonntag, C. *The Chemical Basis of Radiation Biology*; Taylor & Francis: London, 1987.
- Tallman, K. A.; Greenberg, M. M. *J. Am. Chem. Soc.* **2001**, *123*, 5181–5187.
- Carter, K. N.; Greenberg, M. M. *J. Am. Chem. Soc.* **2003**, *125*, 13376–13378.
- Miaskiewicz, K.; Osman, R. *J. Am. Chem. Soc.* **1994**, *116*, 232–238.
- Hashimoto, M.; Greenberg, M. M.; Kow, Y. W.; Hwang, J.-T.; Cunningham, R. P. *J. Am. Chem. Soc.* **2001**, *123*, 3161–3162.
- DeMott, M. S.; Beyret, E.; Wong, D.; Bales, B. C.; Hwang, J.-T.; Greenberg, M. M.; Demple, B. *J. Biol. Chem.* **2002**, *277*, 7637–7640.
- Makino, K.; Ide, H. *J. Biol. Chem.* **2003**, *278*, 25264–25272.
- Kroeger, K. M.; Jiang, Y. L.; Kow, Y. W.; Goodman, M. F.; Greenberg, M. M. *Biochemistry* **2004**, *43*, 6723–6733.
- Dizdaroglu, M.; Schulte-Frohlinde, D.; Von Sonntag, C. *Z. Naturforsch.* **1977**, *32c*, 1021–1022.
- Goyne, T. E.; Sigman, D. S. *J. Am. Chem. Soc.* **1987**, *109*, 2846–2848.
- Hwang, J.-T.; Tallman, K. A.; Greenberg, M. M. *Nucleic Acids Res.* **1999**, *27*, 3805–3810.
- Zheng, Y.; Sheppard, T. L. *Chem. Res. Toxicol.* **2004**, *17*, 197–207.
- Lue, R. Y. P.; Chen, G. Y. J.; Hu, Y.; Zhu, Q.; Yao, S. Q. *J. Am. Chem. Soc.* **2004**, *126*, 1055–1062.
- See Supporting Information.
- Kim, J.; Gil, J. M.; Greenberg, M. M. *Angew. Chem., Int. Ed.* **2003**, *42*, 5882–5885.
- Shibutani, S.; Takeshita, M.; Grollman, A. P. *J. Biol. Chem.* **1997**, *272*, 13916–13922.
- Stubbe, J.; Kozarich, J. W. *Chem. Rev.* **1987**, *87*, 1107–1136.
- Lindahl, T. *Nature* **1993**, *362*, 709–715.

JA0426185